Cobalt Chloride Equilibrium Lab

Name:	
My Mark:/9	
Teacher Mark:/9	

Follow the directions on Station C of your lab for the following equilibrium:

$$Co(H_2O)_6^{2+} + 4 Cl^- \longleftrightarrow CoCl_4^{2-} + 6 H_2O$$

Observations:

- 1) What was the colour of the solution before adding stress? (1 mark)
- 2) Complete the following chart. The first two rows will be answered in your group. You will answer the rest based on a demonstration in class. (8 marks)

Stress	Colour of the Solution	Why?
Heating		
Cooling		
Adding HCl		
Adding AgNO ₃ (silver bonds with Cl ⁻ to form AgCl)		